

Simulation analysis of the effects of aerosol and surface albedo on the remote sensing detection of greenhouse gases in the short-wave infrared 1.6 μm

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Abstract: Aerosols and surface albedo are major sources of error in retrieving greenhouse gas concentrations using high-resolution shortwave infrared spectroscopy. This study employs a high-precision atmospheric radiative transfer model to simulate the influence of aerosols and six different surface types on satellite-observed spectra in the 1594 nm~1624 nm and 1662 nm~1672 nm bands. The results indicate that as aerosol optical depth (AOD) increases, radiance generally increases, with the most significant change observed over vegetated surfaces, which show a 13.26% variation. Within the CO₂ and CH₄ absorption bands, the increments of CO₂/CH₄ under equivalent radiation corresponding to the six surface types are ranked: vegetation, metal material, building material, sedimentary finerock, soil, and sedimentary coarserock. Taking soil surface as an example, the study finds that radiance decreases by approximately 0.41 W/m²/μm/sr for every 1 ppm increase in CO₂ concentration and by about 0.86 W/m²/μm/sr for every 1 ppb increase in CH₄ concentration. Further analysis shows a near-parabolic relationship between AOD and radiance, with consistent trends for CO₂ and CH₄. As AOD increases, the concentration of both gases exhibit continuous growth. Vegetated surfaces demonstrate the largest concentration changes, with CO₂ and CH₄ varying by approximately 40.96 ppm and 137.87 ppb, respectively. Explorations under mixed surface conditions indicate that spectral radiance increases with surface albedo, reaching maximum values of 7.7 W/m²/μm/sr for CO₂ and 7.45 W/m²/μm/sr for CH₄. These findings underscore the critical roles of aerosols and surface albedo in satellite-based greenhouse gas retrievals, offering valuable theoretical guidance for enhancing the accuracy of remote sensing measurements.

Keywords: Aerosol and surface; Remote sensing detection; Greenhouse gases.

1. Introduction

Since the industrial revolution, intensified human activities—particularly the extensive use of fossil fuels such as coal, oil, and natural gas—have disrupted the natural balance among the atmosphere, oceans, and biosphere, leading to a significant rise in greenhouse gas (GHG) concentrations [[1]]. According to the Greenhouse Gas Bulletin published by the World Meteorological Organization (WMO) in 2023, carbon dioxide (CO₂) and methane (CH₄), the two principal atmospheric greenhouse gases, account for approximately 82% of the greenhouse effect. From 1750 to 2023, the global average of CO₂ concentration increased from 280 ppm to 420 ppm, nearly a 1.5-fold rise compared to pre-industrial levels, while the global average CH₄ concentration surged to 2.6 times its pre-industrial level. As of 2023, the global average near-surface temperature has risen by approximately 1.45°C above pre-industrial levels, with China's average temperature increasing by 0.81°C since 1991 [[2]]. Over the past 70 years, China's temperature has shown a particularly pronounced upward trend, warming at an average rate of 0.3°C per decade—nearly double the global average—largely attributed to rapid industrialization. The escalating concentrations of greenhouse gases have not only

accelerated global warming but have also amplified the frequency and intensity of extreme weather events, exerting profound impacts on human societies' survival and lifestyles. These challenges underscore the urgency of addressing greenhouse gas emissions and mitigating the far-reaching consequences of climate change [[3],[4]].

There are three main methods of GHG detection: ground-based, airborne and satellite detection. Ground-based detection is the earliest and has established observation system such as the Total Carbon Column Observation Network (TCCON) [[5]] and the Network for Detection of Atmospheric Composition Changes (NDACC) [[6]]. On the airborne front, a series of detectors have also been developed, including the Airborne Laser Infrared Absorption Spectrometer (ALIAS) [[7]] and the second-generation ALIAS (ALIAS-II) [[8]], the Airborne Visible/Infrared Imaging Spectrometer (AVIRIS-Classic) [[9]] and the next-generation AVIRIS (AVIRIS-NG) [[10]], and the Unmanned Aerial Systems Chromatography for Atmospheric Trace Matter Sounding (UCATS) [[11]], etc. While ground-based and airborne observations are highly accurate, they have limited spatial coverage and can only be applied to localized observations. In contrast, satellite observation is not constrained by geographic or natural conditions, enabling sustained and high-precision global monitoring. Satellite

measurements offer valuable “top-down” validation for “bottom-up” emission inventories, supporting large-scale, accurate greenhouse gas monitoring [[12],[13],[14]]. Currently, major satellites dedicated to greenhouse gas monitoring include Sentinel-5p satellite [[15]], which enables global monitoring of CO₂ and CH₄ concentrations via the TROPOMI payload, and satellites in the GOSAT and OCO series, along with point-source monitoring satellites like GHGsats, Methasat and CarbonMapper. Since China launched its first carbon monitoring satellite, TanSat [[16]], in 2016, it has also launched a series of satellites such as GF-5 [[17]], FY-3D [[18]] and DQ-1 [[19]], many of which involve Polarization Cross Fire (PCF) observation modes.

A key challenge in satellite-based GHG monitoring is the selection of appropriate detection bands. Early remote sensing detection of CO₂ concentration mainly relies on the thermal infrared band, which is sensitive to CO₂ concentration in the middle and upper layers but has a large uncertainty in the detection of gas concentration in the bottom layer [[20]], and the information on greenhouse gas concentration can be performed from the data of some comprehensive remote sensing payloads configured with this band, which can satisfy the demand for early greenhouse gas detection. At present, the new generation of on-orbit GHG detection satellite-based passive remote sensing payloads have chosen near-infrared (NIR) bands to realize the detection of GHG concentrations in the lower atmosphere (troposphere and boundary layer) [[21]], and the selected NIR bands are [[22],[23]]: the 1.56 μm CO₂ absorption band and the 1.64 μm CH₄ absorption bands, which are sensitive to CO₂ and CH₄ concentration variations and have few interfering gases, respectively, and are used for inversion of CO₂ and CH₄. In the shortwave infrared region, there are many generalized and combined bands for CO₂ and CH₄ molecules, especially near 1.6 μm. Using this band, CO₂ and CH₄ absorption features can be effectively characterized to obtain CO₂ and CH₄ concentration information. Moreover, short-wave infrared is more sensitive to the near-surface CO₂ and CH₄ concentration changes, which can make up for the shortcomings of thermal infrared [[4],[24]].

The primary requirement for accurate greenhouse gas detection using the solar short-wave infrared band is simulating the solar spectrum through an atmospheric radiative transfer model. The model simulates the value of radiation received by satellite sensors after the solar spectrum has been absorbed, scattered and reflected by the Earth's surface. Satellites measure the energy backscattered by the atmosphere and reflected from the ground, but the scattering of light by aerosols introduces changes in photon behavior within the aerosol layer, altering the optical path length. As a result, aerosol optical properties and surface characteristics are among the most important sources of error in satellite measurements [[25]]. Aerosol scattering can extend the optical path length, potentially causing overestimation of gas concentrations if multiple scattering effects are not properly accounted for in inversion algorithms. Conversely, neglecting the shortening of the optical path length due to backward scattering may lead to underestimation of the actual concentrations [[26]]. The reflectance spectra observed by satellites are highly dependent on surface albedo. Significant variations in the albedo of different surface types result in pronounced differences in the absorption spectra detected by satellites. Surface albedo is therefore a critical factor influencing the accuracy of atmospheric CO₂ and CH₄

concentration measurements. Consequently, understanding the impact of different aerosol properties and surface characteristics on the satellite-observed CO₂ and CH₄ absorption spectra is essential for accurately retrieving CO₂ and CH₄ concentrations.

In summary, this study investigates the influence of varying aerosol optical thicknesses and surface types on satellite-observed radiation, utilizing the LINTRAN radiative transfer model in accordance with the technical specifications of greenhouse gas hyperspectral detection instruments. This approach offers a robust theoretical foundation for analyzing remote sensing error sources in atmospheric CO₂ and CH₄ retrievals via short-wave infrared spectroscopy. Furthermore, it provides valuable technical support for the development of high-precision algorithms for retrieving CO₂ and CH₄ column concentrations.

2. Methods and Principles

2.1. Atmospheric Radiative Transfer Model

The forward modeling of remote sensing by atmospheric satellites characterizes the radiative transfer process, which translates atmospheric and surface variables—primarily including micro-parameters such as aerosols, atmospheric molecules, and greenhouse gases—into the radiance observed at the top of the atmosphere by the satellite. This process can be mathematically expressed as:

$$y = F(x, b) + e_y \quad (1)$$

where y is the measurement vector, x is the state vector, b containing ancillary parameters, e_y represents the measurement error vector, and F denotes the forward model. The measurement vector consists mainly of multi-band intensity and polarization observation information [[27]], which has the Stokes parameters I , Q , U and V [[28]]. as its components:

$$I = [I, Q, U, V]^T \quad (2)$$

the Stokes parameters are calculated by the SRON radiative transfer model LINTRAN v2.0 [[29]]. Using linear polarization degree ($DoLP$) to describe polarization observation, the calculation is as follows:

$$DoLP = \frac{\sqrt{Q^2 + U^2}}{I} \quad (3)$$

In this study, the forward model integrates atmospheric components (greenhouse gases and aerosols), surface characteristics, and vector radiative transfer to simulate the coupling effects between these factors. The aerosol model is categorized into three distinct modes, with the particle size distribution of each mode characterized by a log-normal distribution. Mode one represents fine particles, while modes two and three correspond to insoluble coarse particles and soluble coarse particles, respectively. Mode one primarily consists of inorganic aerosols, black carbon aerosols, and organic carbon aerosols. Mode two is dominated by non-spherical components such as sand and dust, and mode three is mainly composed of inorganic aerosols. The aerosol parameters considered in the model include aerosol optical depth (AOD), effective radius (r_{eff}), effective variance (v_{eff}), sphere fraction (f_{sph}), aerosol layer height (z_{acr}) and the refractive index coefficients corresponding to the standard spectra of each modal component [[30]].

To describe the surface, the surface reflection matrix $R_{surf}(\lambda, \theta_s, \theta_v, \varphi)$ is given by

$$R_{\text{surf}}(\lambda, \theta_s, \theta_v, \varphi) = r_{11}(\lambda, \theta_s, \theta_v, \varphi)D + R_{\text{pol}}(\theta_s, \theta_v, \varphi) \quad (4)$$

where λ is the wavelength; θ_s and θ_v are the sun zenith angle and view zenith angle, respectively; φ is the relative azimuth angle; D is a null matrix except $D_{11} = 1$, $r_{11}(\lambda, \theta_s, \theta_v, \varphi)$ is

$$r_{11}(\lambda, \theta_s, \theta_v, \varphi) = A(\lambda)(1 + k_{\text{geo}}f_{\text{geo}}(\theta_s, \theta_v, \varphi) + k_{\text{vol}}f_{\text{vol}}(\theta_s, \theta_v, \varphi)) \quad (5)$$

where $A(\lambda)$ is the isotropic reflectance; $f_{\text{geo}}(\theta_s, \theta_v, \varphi)$ and $f_{\text{vol}}(\theta_s, \theta_v, \varphi)$ are the geometric (Li-Sparse) kernel and volumetric (Ross-Thick) kernel function, respectively; and k_{geo} and k_{vol}

described by the Ross–Li bi-directional reflection distribution function (BRDF) model [[30]],

are the coefficients for the Li-Sparse and Ross-Thick respectively [[31]].

Table 1. Aerosol parameters in the forward models. c1, c2 and c3 correspond to coefficients for standard refractive index of inorganic aerosol (INORG), black carbon (BC), organic carbon (OC) and dust (DU) [[30]].

| Property | | Full name | Parameter Setting |
|-----------------------|---------------------|--------------------------------------|-------------------|
| Fine mode | r_{eff} | Effective radius (mode 1) | 0.15 |
| | v_{eff} | Effective variance (mode 1) | 0.2 |
| | AOD | Aerosol optical depth (mode 1) | 0.05-1.0 |
| | f_{sph} | Spherical fraction (mode 1) | 1.0 |
| | z_{aer} | Aerosol layer height (mode 1) | 2000 |
| | $c_1(\text{INORG})$ | Refractive index coefficient (INORG) | 0.95 |
| | $c_2(\text{BC})$ | Refractive index coefficient (BC) | 0.01 |
| Coarse insoluble mode | $c_3(\text{OC})$ | Refractive index coefficient (OC) | 0.02 |
| | r_{eff} | Effective radius (mode 2) | 1.0 |
| | v_{eff} | Effective variance (mode 2) | 0.6 |
| | AOD | Aerosol optical depth (mode 2) | 0.05-1.0 |
| | f_{sph} | Spherical fraction (mode 2) | 0.0 |
| | z_{aer} | Aerosol layer height (mode 2) | 2000 |
| Coarse soluble mode | $c_1(\text{DUST})$ | Refractive index coefficient (DUST) | 1.0 |
| | r_{eff} | Effective radius (mode 3) | 2.5 |
| | v_{eff} | Effective variance (mode 3) | 0.6 |
| | AOD | Aerosol optical depth (mode 3) | 0.05-1.0 |
| | f_{sph} | Spherical fraction (mode 3) | 1.0 |
| | z_{aer} | Aerosol layer height (mode 3) | 500 |
| | $c_1(\text{INORG})$ | Refractive index coefficient (INORG) | 0.9 |

For the surface types, six surface types were considered: soil, vegetation, sedimentary finerock, sedimentary coarserock, building material, metal material. Except for vegetation, the reflectance spectra of other surface types lack distinct peaks and valleys. The reflectance of soil is influenced by factors such as particle size, water content, and organic matter content. Rock reflectance spectra are primarily affected by mineral composition, degree of weathering, and other geological factors. Both soil and sedimentary coarserock exhibit water absorption features at 1.4 μm and 1.9 μm . Man-made surfaces were categorized into two types: building material and metal material. Building material, including concrete and asphalt, typically exhibit low and stable reflectance values around 0.3, with a relatively flat spectral curve. Metallic material, on the other hand, have even lower reflectance, with their spectral characteristics largely determined by surface roughness and the degree of oxidation. The reflectance spectrum of the vegetation follows a unique pattern that can be divided into three distinct regions. In the visible band, vegetation exhibits a small reflectance peak at approximately 0.55 μm , flanked by absorption features at 0.35 μm and 0.67 μm , caused by chlorophyll absorption of

blue and red light while reflecting green light. In the near-infrared band, vegetation reflectance shows a characteristic “steep slope” with a peak near 1.1 μm , creating a distinct spectral signature.

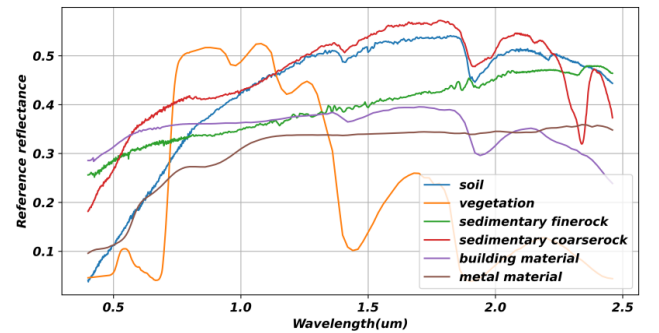


Fig. 1 Reference reflectance for soil, vegetation, sedimentary finerock, sedimentary coarserock, building material and metal material. The data were downloaded from ASTER spectral library.

In the standard setup, the model atmosphere is discretized into 20 homogeneous vertical layers. The vertical profile data for CO_2 and CH_4 were obtained from the American Standard Atmospheric Profile Information [[32]]. Basic atmospheric

parameters, including temperature, humidity and pressure, were read from the NCEP reanalysis data. The absorption coefficients and diffusion characteristics of greenhouse gases at various wavelengths and spectral resolutions were calculated using the Absorption Cross Section and HighResolution Transmittance Database HITRAN [[33]]. The solar zenith angle (SZA) was set to 60°. In this study, the forward model mainly consists of an atmospheric model (greenhouse gases, aerosols) and a surface model, as well as a vector radiative transfer that couples them. The overall framework of the forward model is outlined below:

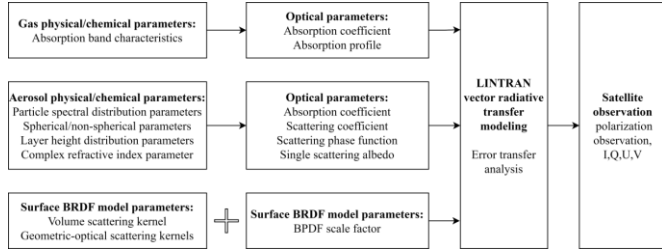


Fig.2 Forward Modeling Flowchart.

2.2. Assessment methodology

This study primarily examined the influence of varying AOD and surface albedo on the radiance of CO₂ and CH₄. The rate of change in radiance under these conditions is expressed by the following equation [[34]]:

$$R_change_n = \frac{R_n - R_0}{R_0} \times 100\% \quad (6)$$

According to the 20th Global Greenhouse Gases Bulletin published by the World Meteorological Organization [[35],[36]], the average annual increase in concentration of CO₂ is 2.3 ppm, while the average annual increase in concentration of CH₄ is 11 ppb. In this study, the average concentration of CO₂ and CH₄ provided in the bulletin was used as the initial value, and the annual increase was taken as the magnitude of the change in concentration. The initial CO₂ concentration was set to 420 ppm, with simulations conducted under clear-sky conditions. The CO₂ concentration was subsequently adjusted to 422.3 ppm, reflecting an increase of 2.3 ppm. Similarly, the initial CH₄ concentration was set to 1934 ppb, with an adjustment of 11 ppb. The average radiance change corresponding to a 1 ppm increase in CO₂ (or a 1 ppb increase in CH₄) was calculated using the following formula, where \bar{R} indicates the average value of radiance. For CO₂, *wave* refers to the spectral ranges 1601 nm~1604 nm and 1607 nm~1611 nm, while for CH₄, *wave* corresponds to 1665 nm~1667 nm.

$$\bar{R}(\Delta 1 ppm) = \frac{[\bar{R}_{wave}(422.3 ppm) - \bar{R}_{wave}(420 ppm)]}{2.3 ppm} \quad (7)$$

$$\bar{R}(\Delta 1 ppb) = \frac{[\bar{R}_{wave}(1945 ppb) - \bar{R}_{wave}(1934 ppb)]}{11 ppb}$$

The CO₂ concentration is determined by dividing the average radiance value within the high-sensitivity band by the radiance value corresponding to a 1 ppm change under identical conditions. The calculation method for CH₄ follows the same approach as that for CO₂.

$$\Delta CO_{2(n)} = \frac{(\bar{R}_{wave(n)} - \bar{R}_{wave(n-1)})}{\bar{R}(\Delta 1 ppm)} \quad (8)$$

$$\Delta CH_{4(n)} = \frac{(\bar{R}_{wave(n)} - \bar{R}_{wave(n-1)})}{\bar{R}(\Delta 1 ppb)}$$

3. Sensitivity Experiment

3.1. Initial Sensitivity Analysis of CO₂ and CH₄ Concentration

In this study, an atmospheric radiative transfer model was employed to examine the impact of a 1 ppm or 1 ppb change in CO₂ and CH₄ concentrations on the spectral radiation. The spectral resolution used in the simulation was 0.12 nm, with the wavelength range was 1594 nm ~1624 nm for CO₂ and 1662 nm ~1672 nm for CH₄. For demonstration purposes, soil was selected as the surface type, and the AOD was set to 0.1. When the initial CO₂ concentration was adjusted by 2.3 ppm, the rate of change in radiance ranged from -0.16% to 0%. The bands near 1594 nm~1596 nm showed the smallest rate of change in radiance due to the change in CO₂ concentration, indicating that these bands are not sensitive to the change in CO₂ concentration. On the other hand, the maximum sensitivity of -0.16% for the bands near 1601 nm~1604 nm indicates that these bands are very sensitive to changes in CO₂ concentration and that these absorption peaks are relatively densely located. Since the depth of the absorption peaks varies between wavelengths, in order to facilitate the adjustment of the calculation of the initial CO₂ value, the interval of -0.1% to -0.16% change in radiance brightness was selected as the highly sensitive bands for CO₂, and about 58 specific bands in the ranges of 1601 nm~1604 nm and 1607 nm~1611 nm were finally identified, and their average values were calculated, and it was found that for every 1 ppm increase in the concentration of CO₂, the radiance decreases by about 0.406063 W/m²/um/sr over these 58 specific bands($\bar{R}(\Delta 1 ppm)$).

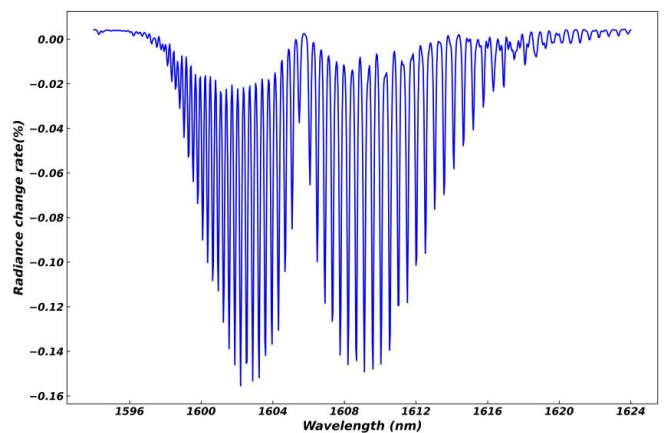


Fig.3 Radiance change rate for CO₂ with an initial estimate of 420 ppm and a variation of 2.3 ppm for AOD@550nm is 0.1 and for soil surface.

When the initial value of CH₄ is adjusted by 11 ppb, the range of radiance change rate is from -0.21% to 0%. The depth of the absorption peaks of CH₄ is not as intensive as that of CO₂, which is mainly manifested near the 1666 nm band. In order to facilitate the adjustment of the calculation of the initial CH₄ value, the radiance change rate of -0.1% to -0.22% interval was selected as the highly sensitive bands of CH₄, and

about 21 specific bands in the range of 1665 nm~1667 nm were finally identified and their average values were calculated, and it was found that for every increase of 1 ppb in the concentration of CH₄, the radiance decreases by about 0.861274 W/m²/um/sr($\bar{R}(\Delta 1\text{ppb})$).

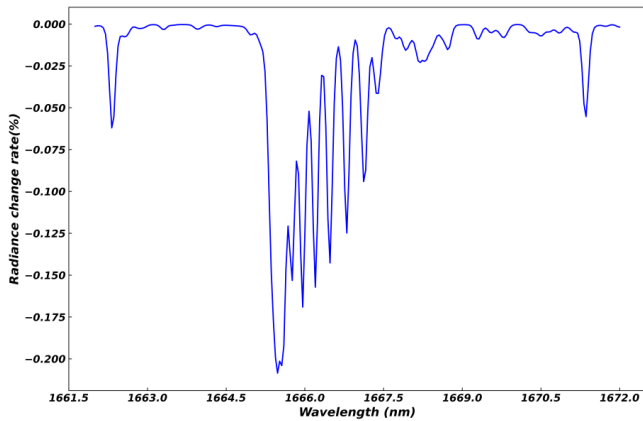
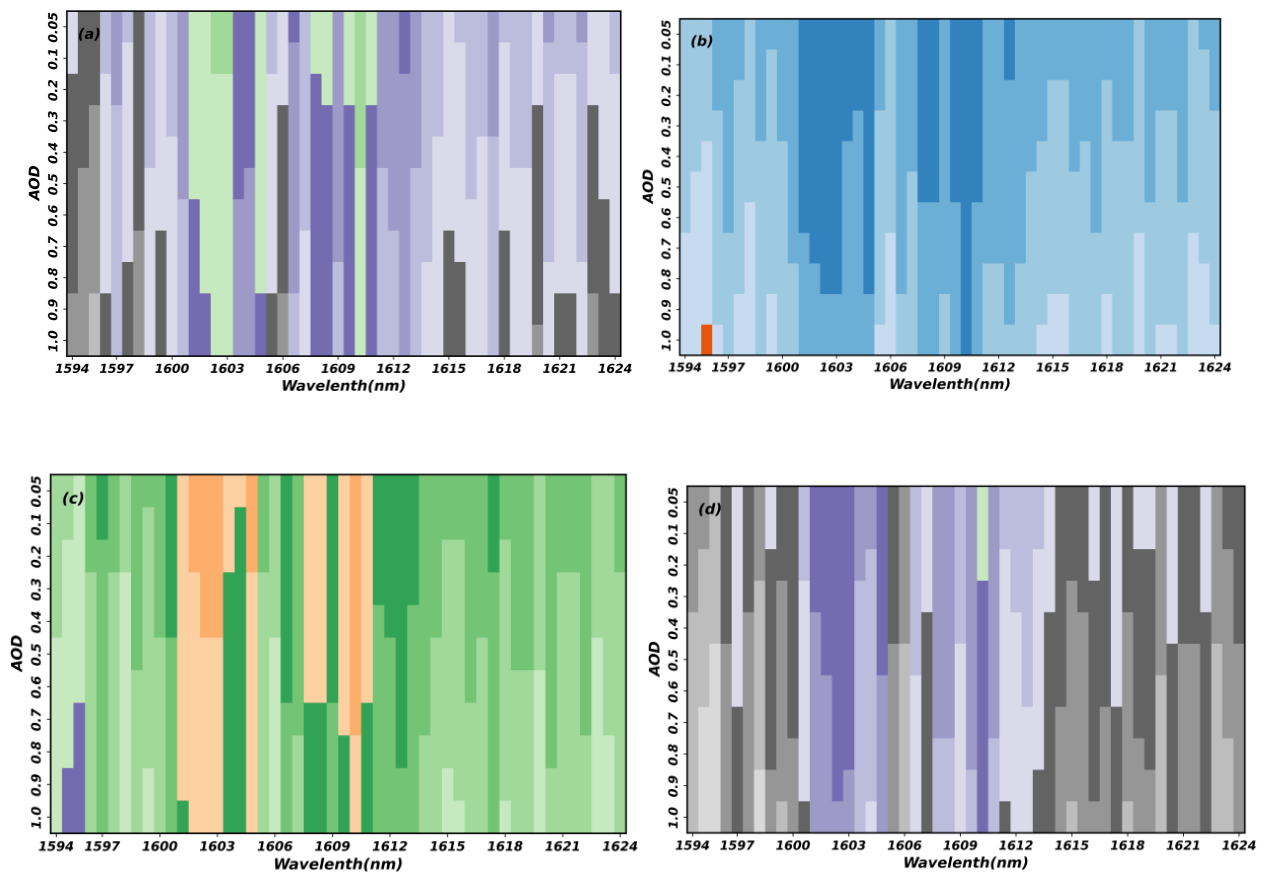


Fig. 4 Radiance change rate for CH₄ with an initial estimate of 1934 ppb and a variation of 11 ppb for AOD@550nm is 0.1 and for soil surface.

3.2. Influence of aerosol and surface albedo on CO₂, CH₄ radiance

The figure below illustrates the variations in CO₂ radiance across different surface types under varying AOD. As the AOD increases from 0.05 to 1.0, a consistent upward trend in radiance is observed. This rise in AOD corresponds to a higher concentration of aerosol particles in the atmosphere, which significantly enhances their scattering and absorption of light, thereby increasing atmospheric radiance. Notable fluctuations are evident under varying AOD conditions. Among the surface types, sedimentary coarserock and soil exhibits the most pronounced radiance changes, transitioning across three distinct color ranges (gray-purple-green) as AOD varies. In contrast, vegetation predominantly appears in blue, indicating minimal radiance variation. The radiance changes of sedimentary finerock and building material are nearly identical due to their similar reflectance properties, with green and orange dominating their color spectrum. In the CO₂ strong absorption bands (1601 nm~1604 nm and 1607 nm~1611 nm), sedimentary coarserock surfaces exhibit the highest radiance values, followed by soil, sedimentary finerock, building material, metal material, and vegetation in descending order. This ranking aligns with the surface reflectance characteristics of these materials in the 1.6 μm band.



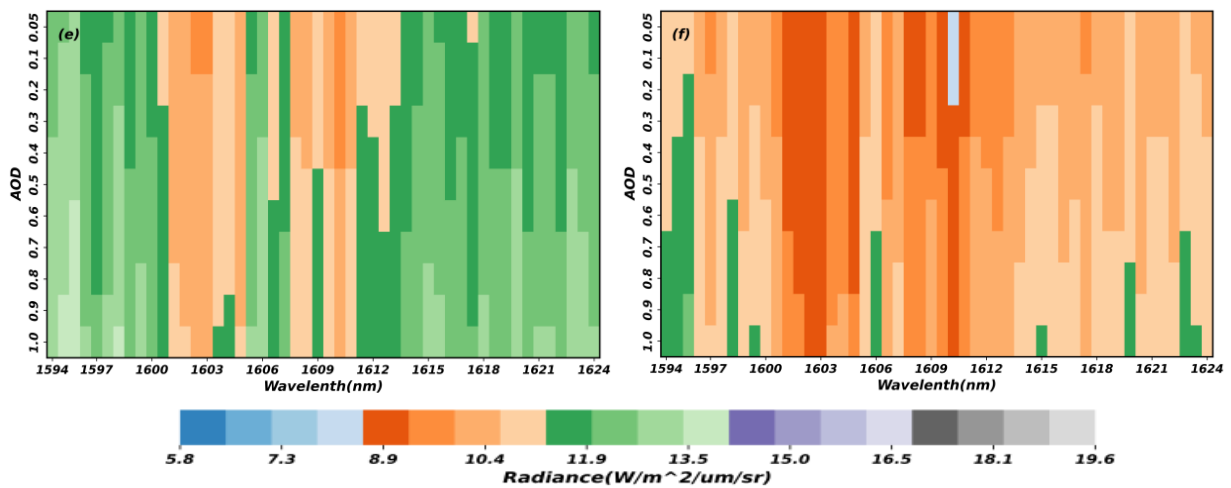
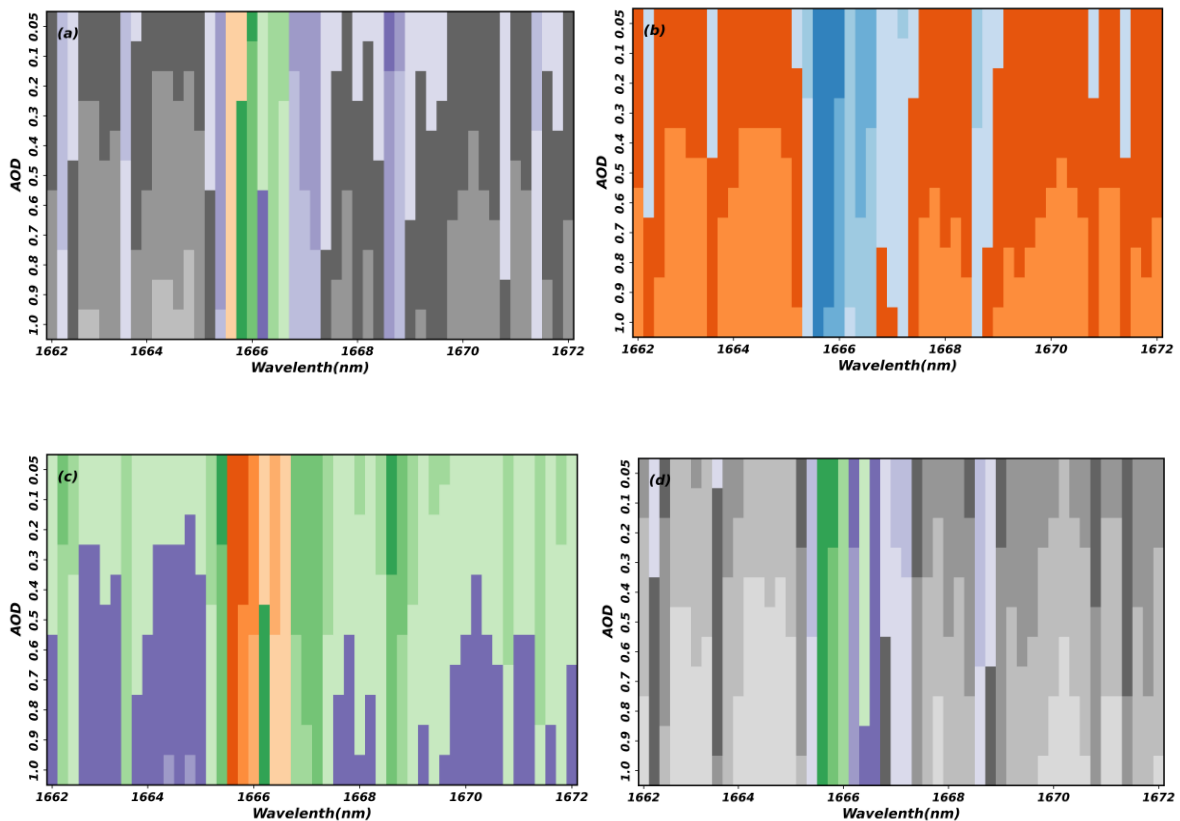


Fig. 5 CO₂ Radiance at different AOD for different surface types. (a)soil; (b)vegetation; (c)sedimentary finerock; (d)sedimentary coarserock; (e)building material; (f)metal material

Similar to CO₂, the radiance of CH₄ also increases progressively with the rise in AOD. As shown in the figure, CH₄ radiance is notably higher in the weak absorption bands (1662.5 nm~1665 nm and 1667 nm~1672 nm) and lower in the strong absorption band (1665 nm~1667 nm). Among surface types, soil and sedimentary coarserock exhibit the most pronounced variations in radiance, appearing predominantly gray and purple in the weak absorption bands and green in the strong absorption band. In contrast, vegetation shows the least variation, with orange dominating the weak absorption bands and blue prevailing in the strong absorption band. Within the CH₄ strong absorption

band, sedimentary coarserock surface display the highest radiance values, followed in descending order by soil, sedimentary finerock, building material, metal material, and vegetation. In summary, the combined effects of surface albedo and AOD govern the variations in atmospheric radiance. Under high-albedo conditions, increased surface reflectance significantly amplifies aerosol scattering, resulting in a notable rise in atmospheric radiance, as observed over soil and sedimentary coarserock. These findings reveal the synergistic mechanisms between aerosols and surface albedo in shaping atmospheric radiative characteristics.



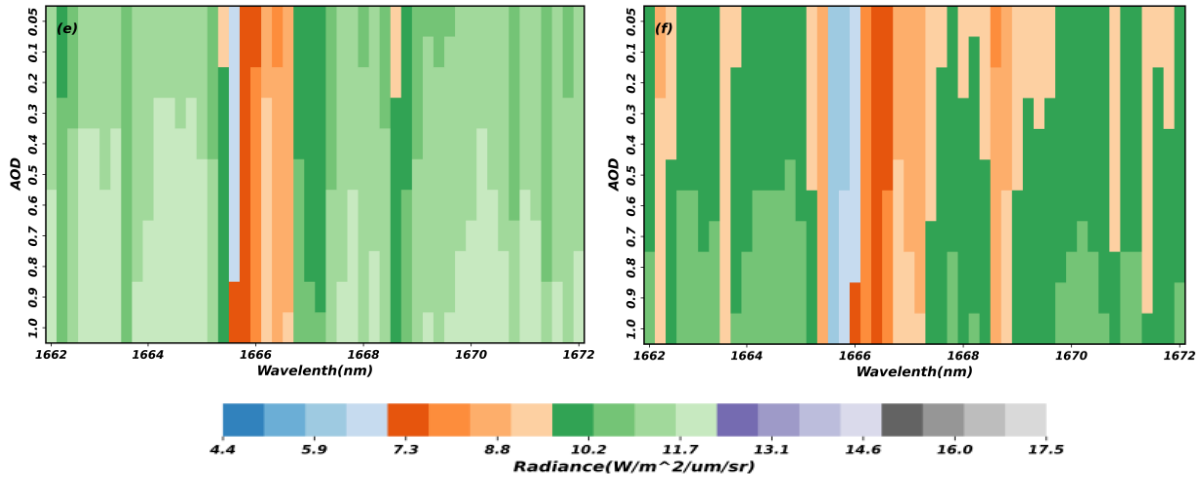


Fig. 6 CH₄ Radiance at different AOD for different surface types. (a)soil; (b)vegetation; (c)sedimentary finerock; (d)sedimentary coarserock; (e)building material; (f)metal material

The rate of change of CO₂ and CH₄ radiance at varying AOD levels was calculated using the AOD-neutral case as a standard reference, with radiance differences derived from the highly sensitive bands identified in Section 3.1. The results are shown below. By comparing the spectral changes induced by each 1 ppm or 1 ppb increase in CO₂ and CH₄ concentrations, a relationship between AOD and changes in CO₂ and CH₄ concentrations was established. The study reveals a consistent trend between CO₂ and CH₄ concentrations. With the increase of AOD, the concentrations of both gases show a continuous growth trend. However, when AOD reaches a high value, the rate of concentration growth begins to slow down. This finding aligns with findings from the SRON team in the Netherlands on CO₂ inversion for the CO2M satellite, which noted that for fine aerosols in the boundary layer, the XCO₂ bias is estimated to be limited but still slightly grows for larger AOD. Additionally, for coarse

aerosols around 8 km altitude, such as dust and thin cirrus, the XCO₂ bias is also dependent on AOD but is estimated to be several ppm [[37]]. These results corroborate our forward modeling process, where the gas concentration growth rate similarly diminishes at higher AOD values. This nuanced relationship underscores the complex interplay between AOD and gas concentration, highlighting the existence of a critical AOD level that balances the competing effects of aerosols on gas detection. Among the various surface types, the variations of approximately 40.96 ppm and 137.87 ppb, respectively. The magnitude of changes in gas concentrations across surface types ranked in descending order is as follows: vegetation, metal material, building material, sedimentary finerock, soil, and sedimentary coarserock. Due to the similar surface albedo of soil and coarserock, their AOD-induced changes in CO₂ and CH₄ concentrations exhibited nearly identical trends, resulting in overlapping curves.

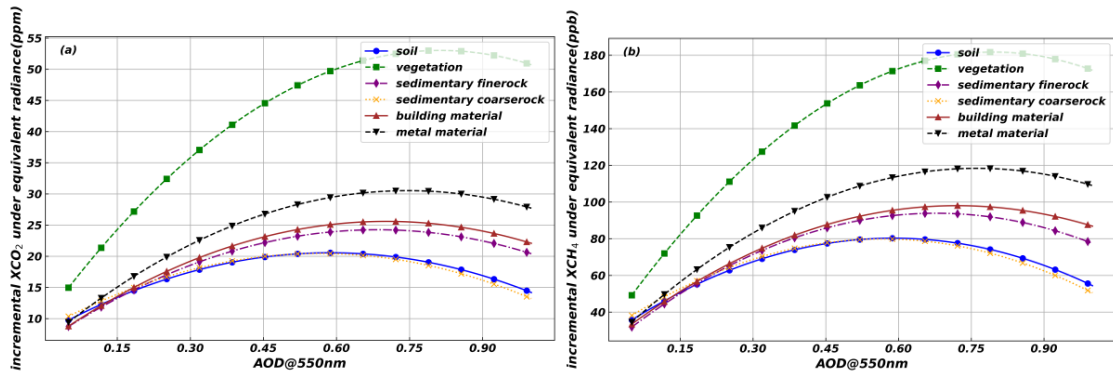


Fig. 7 Incremental XCO₂/XCH₄ under equivalent radiance for AOD@550nm. (a)XCO₂; (b)XCH₄(XCO₂ and XCH₄ are calculated from Eq. (8).)

Table 2. The equation coefficients between AOD@550nm and gas concentration changes.

| surface type | AOD@550nm & CO ₂ | | | AOD@550nm & CH ₄ | | |
|------------------------|-----------------------------|----------|----------|-----------------------------|----------|----------|
| | <i>a</i> | <i>b</i> | <i>c</i> | <i>a</i> | <i>b</i> | <i>c</i> |
| soil | -37.44 | 43.98 | 7.64 | -153.36 | 180.72 | 27.02 |
| vegetation | -65.42 | 106.35 | 9.79 | -238.36 | 379.39 | 30.76 |
| sedimentary finerock | -38.58 | 52.81 | 6.18 | -157.67 | 213.59 | 21.52 |
| sedimentary coarserock | -38.18 | 43.10 | 8.29 | -156.25 | 177.06 | 29.68 |
| building material | -39.53 | 55.50 | 6.11 | -143.54 | 206.93 | 23.41 |
| matal material | -43.59 | 65.02 | 6.27 | -165.61 | 252.21 | 22.34 |

The variation of AOD demonstrates an almost parabolic relationship with the concentration changes of CO₂ and CH₄,

with the fitting curve following a quadratic form. The equations are expressed as $y=ax^2+bx+c$, where y represents

the gas concentration, a , b and c are coefficients derived from the fitting process, and x denotes the AOD. The curve equation coefficients for the six surface types are shown in the table below. With the exception of vegetation, the gas concentration values for all other surface types remain nearly constant and show minimal variation when the AOD is 0.05.

3.3. Simulation and analysis of CO₂ and CH₄ based on mixed surface

In this study, the variations in CO₂ and CH₄ radiance across six different surface types were analyzed, with a particular focus on the impact of mixed surfaces on the remote sensing

detection of greenhouse gases in the short-wave infrared 1.6 μm [[38]]. The mixed surface simulation involved a random mixture of soil and vegetation, with aerosol parameters and SZA settings consistent with those used in previous studies (see Section 2). Additionally, the CH₄ concentration was fixed at 1934 ppb, and the CO₂ concentration was set at 420 ppm. As shown in the figure, the figure demonstrates the influence of CO₂ and CH₄ radiance brightness in the absorbing band by the magnitude of AOD. The CO₂ and CH₄ radiance gradually increases with increasing AOD. In the strong absorption bands of CO₂ and CH₄, the radiance varies more gently and does not show significant fluctuations.

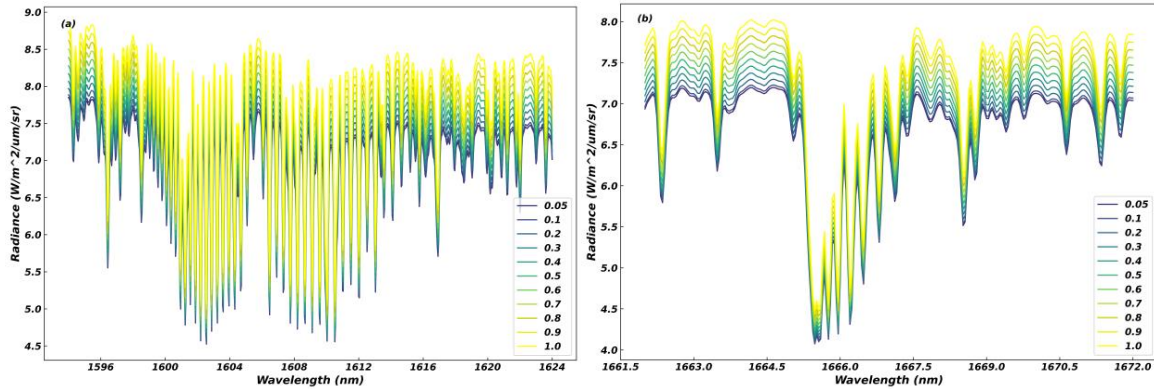


Fig. 8 Radiance for AOD@550nm range from 0.05 to 1.0 and mixed surface. (a)CO₂; (b)CH₄

By grouping and calculating the radiance under mixed surface conditions, the average radiance values is calculated for various combinations of AOD and Albedo. As illustrated in the figure, these combinations have a pronounced effect on the radiance of CO₂ and CH₄. The analysis reveals that both CO₂ and CH₄ radiances generally increase with higher AOD and albedo, reaching maximum average values of 7.7

W/m²/μm/sr and 7.45 W/m²/μm/sr, respectively. At low AOD values, variations in surface albedo exert minimal influence on radiance, as indicated by the deep blue regions in the figure. However, CO₂ radiance is more sensitive to changes in AOD compared to CH₄, with radiance transitioning to a yellow hue as AOD increases to 1.0. This highlights the stronger dependence of CO₂ radiance on atmospheric aerosol content.

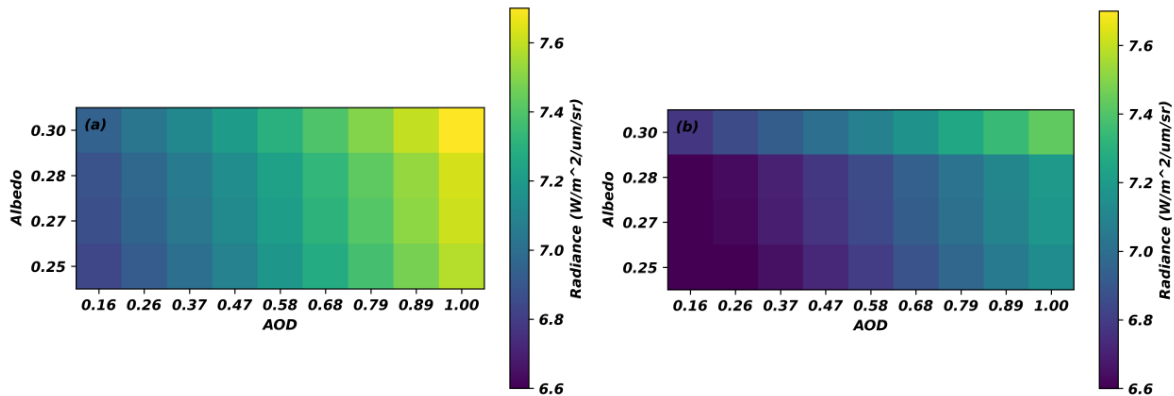


Fig. 9 The average radiance over 1594nm~1624nm and 1662nm~1672nm for different AOD@550nm and albedo for mixed surface. (a)CO₂ is 420ppm; (b)CH₄ is 1934ppb

4. Conclusion

This article quantitatively analyzes the effects of AOD and surface albedo on atmospheric CO₂ and CH₄ detection using the LINTRAN radiative transfer model. First, the concentration changes of CO₂ and CH₄ were calculated to determine the corresponding radiance values. A sensitivity analysis of six surface types and AOD parameters was established to evaluate their impact on the concentration values of CO₂ and CH₄. Subsequently, the relationship between radiance, AOD, and albedo was explored under mixed surface conditions. The main research conclusions are as follows:

As the aerosol optical depth (AOD) increases, the radiance

generally shows an upward trend. The radiance change is most pronounced for the vegetation surface, with a change rate of 13.26%. In the absorption bands of CO₂ and CH₄, the six surface types are ranked in descending order of radiance variation as follows: vegetation, metal material, building material, sedimentary finerock, soil and sedimentary coarserock. Represented by bare soil surface, research has found that for every 1ppm increase in CO₂ concentration, the radiance decreases by approximately 0.41W/m²/um/sr; For every 1 ppb increase in CH₄ concentration, the radiance decreases by approximately 0.86 W/m²/um/sr. Further analysis showed that the changes of AOD and gas concentration were nearly parabolic, and the trends of CO₂ and CH₄ were consistent. With the increase of AOD, the

concentration of both gases showed a continuous increase. Under vegetation surface types, the changes in CO₂ and CH₄ concentrations are most significant, with variations of approximately 40.96 ppm and 137.87 ppb, respectively. Further exploration based on mixed surface conditions shows that as the surface albedo increases, the observed spectral radiance also increases, with the maximum radiance values of CO₂ and CH₄ reaching 7.7 W/m²/um/sr and 7.45 W/m²/um/sr, respectively.

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Disclosures

The authors declare that there are no conflicts of interest related to this paper.

Data availability

We appreciate the information provided by AFGL US standard profile and NCEP reanalysis data. We thank the USGS team for providing the ASTER geophysical spectral database.

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